Vasquez High School -- Chemistry -- Test #1 -- Chapters 1, 2 and Basics -- 100 points

Write TRUE if the statement is true OR write the word that substitutes for the underlined word that would make it true. Writing false only earns partial credit. Three points each.
1) A solid substance that does not form crystals is said to be <u>indefinite</u> .
2) The metric prefix "micro" means one-thousandth.
3) A <u>heterogeneous</u> mixture, like a hot fudge sundae, are different substances throughout it.
4) The boiling point of water in kelvin is 212 K.
5) The metric system is based on the number ten because <u>our number system is too</u> .
6) The way a substance reacts with another is a <u>chemical</u> property.
7) At the freezing point, ALL motion stops.
8) A <u>catalyst</u> speeds up a chemical reaction by lowering the energy necessary to start it.
9) We added <u>sulfuric acid</u> to the sodium hydroxide in demo to neutralize it.
10) I <u>can</u> succeed in this class (and anything else) if I put my sincere effort into it.
Short Answer/Fill-in Section. Write complete answers here. Three points each.
11) The first three steps in the scientific method are:, and
12) Give one example of a qualitative assessment of something:
13) Describe what scientific notation is, why we use it and how we use it:
14) Chemistry is the study of
15) Name three qualities of a scientist from the list I gave you (no making them up!):

a) condensation goes from	16) Give three branches/careers of chemistry AND a definition or description of each:
a) condensation goes from	
b) sublimation goes from	17) Write the changes of phase (from what to what) for these:
c) vaporization goes from	a) condensation goes from
18) Describe the difference between a gas and a vapor: 19) Describe in a few words how the particles of each of these phases of matter behave (in general): a) solid	b) sublimation goes from
Describe in a few words how the particles of each of these phases of matter behave (in general): a) solid	c) vaporization goes from
19) Describe in a few words how the particles of each of these phases of matter behave (in general): a) solid	18) Describe the difference between a gas and a vapor:
b) liquid c) gas 20) How is a compound distinguished from a mixture? Calculation Section. Show all your work here writing the proper equations gets partial credit. Five points. 21) New experiments have determined that the core temperature of the Earth to be about 10832 °F. What is that temperature in °C?	19) Describe in a few words how the particles of each of these phases of matter behave (in general):
c) gas	a) solid
Calculation Section. Show all your work here writing the proper equations gets partial credit. Five points. 21) New experiments have determined that the core temperature of the Earth to be about 10832 °F. What is that temperature in °C?	
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that temperature in °C?	Calculation Section. Show all your work here writing the proper equations gets partial credit. Five points.
22) 5.87 seconds is how many microseconds?	
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23) 216,000,000	0,000 nanoseconds i	s how many seconds?					
25) If there are	25.4 mm in one incl	n, how many centimete	ers are in 500 yards?				
26) The fastest rocket sled could travel at 660 miles per hour. If there are 5280 feet in one mile, how fast is that speed in feet per minute?							
27) For one point each, express these decimal numbers in scientific notation and vice-versa:							
395,000	0.000.0	0000993	2.64 x 10 ⁸	5.17 x 10 ⁻²			
Then express this answer in S/N as well (two points): $(4.8 \times 10^6) \div (8.0 \times 10^{-3})$ Multiple Choice Section. Write the letter that BEST answers each example. Three points each.							
27) Which statement is NOT true?							
a) Pure substances always have the same composition.b) Matter has both physical and chemical properties.c) Elements can be broken down into simpler substances.d) Temperature affects the way particles in a substance are held together.							
28) Whi	ich of the following	is NOT a mixture?					
a) Salt water	b) Coca-cola	c) Wood	d) Carbon dioxide			
29) We wish to calculate the volume of a box. We measure the length to be 25.47 cm, the width to be 18.65 cm, and the height to be 4.6600 cm. The best calculation of volume would have how many significant figures?							
a) Three	b) Four	c) Five	d) Thirteen			